## STUDIES IN SESQUITERPENES—XLVII\* CIS- AND TRANS-ATLANTONES FROM CEDRUS DEODARA LOUD

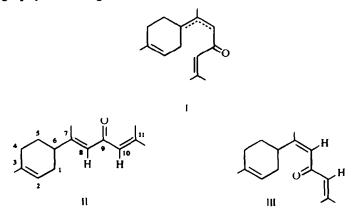
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Abstract—Isolation and characterisation of pure atlantone is reported for the first time, and has been found to have *trans*-geometry. A minor constituent of the essential oil from *Cedrus deodara* Loud is shown to be its *cls*-isomer. Interconversion of the isomers is reported.

ATLANTONES,<sup>1</sup> the major ketones of the essential oil from *Cedrus libanotica* Link., C. atlantica Manet. and C. deodara Loud., have not been obtained pure to date and are considered to be an inseparable mixture of isomers, indicated in formula  $I.^2$ While working<sup>3</sup> on the constituents of *Cedrus deodara*, we became interested in obtaining pure atlantone. We now find that at least the essential oil from *Cedrus* deodara contains essentially one atlantone, which is now shown to have the structure and geometry depicted in II. The only other isomer, which occurs in only minor amounts, is shown to be the *cis*-isomer III. The two isomers can be separated by chromatography on silica gel.



The major isomer<sup>‡</sup> ( $R_f$  0.275) displays in its PMR spectrum four vinylic methyls (1.63, 1.85, 2.10 and 2.10  $\delta$ ), the precise assignments of which are best made<sup>4</sup> in terms of geometry II:  $\delta$  1.63 ( $C_3$ -Me), 1.85 ( $C_{11}$ -trans§-Me), 2.10 ppm ( $C_7$ -Me and  $C_{11}$ -cis§-Me); the 1H multiplet centred at 5.33  $\delta$  is assignable to olefinic proton at  $C_2$ , while the 2H signal (essentially a singlet) at 5.90  $\delta$  is clearly due to the  $C_8$  and  $C_{10}$ 

<sup>‡</sup> The major ketone from the essential oil of *Cedrus deodara* has been earlier identified<sup>3a</sup> as atlantone (I) by its base-catalysed cleavage into acetone and p-methyl- $\Delta^3$ -tetrahydroacetophenone.

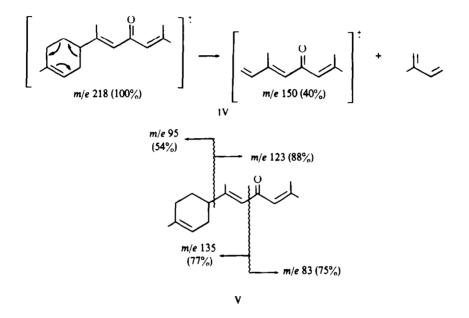
§ Trans- or cis-prefixes for methyls are with respect to the CO group.

<sup>\*</sup> Part XLVI, Tetrahedron 27, 635 (1970).

<sup>†</sup> Communication No. 1490, National Chemical Laboratory, Poona.

protons. On the other hand, the second isomer ( $R_f$  0.365) shows the four vinylic methyls, the chemical shifts of which, are best rationalised in terms of structure and geometry depicted in III:  $\delta$  1.66 broad singlet (C<sub>3</sub>-Me), two doublets centred at 1.80 (J = 2 c/s) and 1.89 with J = 1 c/s (C<sub>7</sub>-Me and C<sub>11</sub>-trans-Me), and 2.16 ppm doublet with J = 1 c/s (C<sub>11</sub>-cis-Me); the three olefinic protons virtually occur at the same field strengths as in the trans-isomer (unresolved 1H multiplet at 5.40  $\delta$ ; 2H multiplet centred at 5.95  $\delta$ ).

The UV (II:  $\lambda_{max}^{EiOH}$  266 mµ,  $\varepsilon$  16,900; III:  $\lambda_{max}^{EiOH}$  266 mµ,  $\varepsilon$  10,210) and IR absorptions (both isomers:  $\nu_{max}$  1670 (low intensity), 1630, 1610 cm<sup>-1</sup>) of both the isomers are as expected.<sup>\*</sup> The two isomers display virtually identical mass spectra; the important fragments are readily rationalised in terms of fragmentations depicted in IV and V.



The cis-isomer (III) is quite unstable and on distillation (bath temp:  $\sim 140^{\circ}$ ) gives a product containing as much as 75–80% trans-isomer (II). As expected,<sup>5</sup> irradiation of trans-isomer results in isomerisation to the cis-isomer.

#### EXPERIMENTAL

UV spectra were taken on a Perkin-Elmer spectrophotometer, model 350. IR spectra were taken on a Perkin-Elmer Infracord model 137E. PMR spectra were measured in 10-20% soln in CCl<sub>4</sub> with TMS as internal standard, on a Varian A-60 spectrometer; signals are recorded in  $\delta$  (ppm) relative to TMS as zero. Mass spectra were measured on a CEC mass spectrometer, model 21-110B, using an ionizing voltage of 70 eV and a direct inlet system.

Silica gel for column chromatography (-100, +200 mesh) was activated at  $125-130^{\circ}/6-8$  hr and standardised according to Hernandez *et al.*<sup>6</sup> TLC was carried out on silica gel layers (0.3 mm) containing 15% gypsum; visualisation of the spots was carried out with 1% vanillin-in-H<sub>3</sub>PO<sub>4</sub> aq (30%) spray, followed by heating at  $\sim 100^{\circ}/10$  min.

\*e.g. cf. phorone<sup>4b</sup>:  $\lambda_{max}^{E1OH}$  265 mµ,  $\epsilon$  19,400,  $\nu_{max}^{CC1_4}$  1678, 1637, 1619 cm<sup>-1</sup>.

GLC was carried out on Aerograph model A-350-B using 300 cm  $\times$  0.6 cm column packed with 20% diethyleneglycol polysuccinate on Chromosorb W (60-80 mesh) with H<sub>2</sub> as carrier gas.

#### Isolation of trans- and cis-atlantones

The essential oil\* (1.84 kg in 20 litre light petroleum, b.p. 40-60°) from the wood of *Cedrus deodara* Loud. was partitioned with 90% aq EtOH (20 l.) in a continuous-type extractor† to finally give a hydrocarbon fraction (520 g) and a hydrocarbon-free oxygenated terpenes cut (400 g), besides a mixture of the two (870 g). The oxygenated terpenes fraction (25 g) was chromatographed (silica gel/IIa, 50 cm  $\times$  5 cms) to get two gross cuts: (I) C<sub>6</sub>H<sub>6</sub> eluate, 500 ml  $\times$  9, 14-68 g; (ii) ether cut, 500 ml  $\times$  6, 9-89 g. The C<sub>6</sub>H<sub>6</sub> eluted material (50 g), which contains atlantone was rechromatographed on silica gel/IIA (70 cm  $\times$  1.6 cm) with TLC monitoring (solvent: C<sub>6</sub>H<sub>6</sub>, two irrigations, 12 cm each time):

Fract. 1	light petroleum 50% $C_6H_6$ in light petroleum $C_6H_6$	500 ml × 2 500 ml × 3 300 ml × 4	n <b>eg</b> ligible.
Fract. 2	00	300 ml × 9	1.6 g, components with $R_f$ 0.4 and 0.32 (trace).
Fract. 3	C <sub>6</sub> H <sub>6</sub>	300 ml × 2	0.735 g, components with $R_f 0.32$ and 0.28.
Fract. 4	C <sub>6</sub> H <sub>6</sub>	300 ml × 5	1.2 g, component with $R_f$ 0.28.
Fract. 5	C <sub>6</sub> H <sub>6</sub> 5% acetone in C <sub>6</sub> H <sub>6</sub>	300 ml × 3 300 ml × 3	
Fract. 6	5% acetone in C <sub>6</sub> H <sub>6</sub>	300 ml × 3	0.277 g, component with $R_f$ 0.18 and 0.16.

Fraction 4, which was essentially atlantone (TLC: single spot; a characteristic green colour. GLC, one major component with retention time 6.4 min; temp 165°; gas: 100 ml/min) resolved into a minor and a major component on TLC with another solvent system (10% CHCl<sub>3</sub> in pet. ether). This fraction (4.3 g) was recharged on silica gel/II (100 cm  $\times$  3.6 cm) using 10% CHCl<sub>3</sub> in light petroleum as the only eluent to finally yield *trans*-isomer II [1.26 g;  $R_f$  0.275, b.p. 142-145° (bath)/1 mm,  $n_{20}^{50}$  1.5342,  $[\alpha]_D$  + 1.2° (c 0.83% in CHCl<sub>3</sub>)] and the cis-isomer III [0.20 g;  $R_f$  0.365,  $n_{20}^{50}$  1.5228]. Mass spectrum (*trans*-isomer): important ions at *m/e* 218 (M<sup>+</sup>, 100%), 203 (M<sup>+</sup>-15, 33%), 150 (40%), 135 (77%), 123 (88%), 107 (61%), 105 (48%), 95 (54%), 91 (50%), 83 (75%), 79 (44%), 77 (42%), 67 (38%) and 55 (45%).

#### Photoisomerisation of trans-atlantone

trans-Atlantone (0.5 g) in heptane (130 ml) was irradiated (100 watts, 'Hanovia' high-pressure, quartz Hg vapour lamp) at room temp with TLC (solvent: 10% CHCl<sub>3</sub> in light petroleum) monitoring. After ~1.5 hr irradiation, sufficient amount of *cis*-isomer had been formed and other compounds started appearing (TLC). At this stage the irradiation was stopped, the solvent flashed off at room temp under vacuum and the product (475 mg) chromatographed on silica gel/II (71 cm × 1.4 cm) using 10% CHCl<sub>3</sub> in light petroleum as eluent to finally give 120 mg pure *cis*-isomer, 100 mg (~90% pure) *trans*-isomer.

#### REFERENCES

<sup>1</sup> A. S. Pfau, Helv. Chim. Acta 15, 1481 (1932)

<sup>2</sup> For an account of earlier chemistry of atlantones see: J. L. Simonsen and D. H. R. Barton, *The Terpenes* Vol. III, pp. 198-202. University Press, Cambridge (1952)

\* Supplied by Standard Essential Oil Distillers, Kanpur (UP).

† The authors are grateful to Dr. M. U. Pai for suggesting the use of this equipment.

- <sup>3</sup> G. S. Krishna Rao, Sukh Dev and P. C. Guha, J. Indian. Chem. Soc. 29, 721 (1952); <sup>b</sup> T. C. Joseph and Sukh Dev, Tetrahedron 24, 3809, 3841, 3853 (1968);

  - <sup>c</sup> R. C. Pandey and Sukh Dev, Ibid. 24, 3829 (1968);
  - <sup>4</sup> S. C. Bisarya and Sukh Dev, Ibid. 24, 3861, 3869 (1968)
- <sup>4</sup> See e.g. <sup>a</sup> L. M. Jackman and S. Sternhell, Applications of Nuclear Magnetic Resonance Spectroscopy in Organic Chemistry pp. 222-225. Pergamon, Oxford (1969); <sup>b</sup> H. N. A. Al-Jallo and E. S. Waight, J. Chem. Soc. (B), 75 (1966)
- <sup>5</sup> See e.g.: M. Mousseron, International Symposium on Organic Photochemistry (IUPAC) pp. 481-492. Butterworths, London (1965)
- <sup>6</sup> R. Hernandez, R. Hernandez Jr. and L. R. Axelrod, Analyt. Chem. 33, 370 (1961)